# Synthesis of Dialkyl 1,4-Dihydro-2,6-dimethyl-4-(heteroaryl)pyridine-3,5-dicarboxylates as Calcium Channel Antagonists Raymond D. Anana and Edward E. Knaus\*

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The Hantzsch condensation of the heteroarylcarboxaldehydes 3a-c with alkyl acetoacetates 4a-c and alkyl 3-aminocrotonates 5a-b afforded the respective dialkyl 1,4-dihydro-2,6-dimethyl-4-(heteroaryl)-pyridine-3,5-dicarboxylates 6a-f possessing a C-4 4-quinolinyl, 8-quinolinyl or 1-oxido-4-pyridinyl substituent. Calcium channel antagonist structure-activity relationships acquired indicate that i) the position of the quinolyl nitrogen atom was not a determinant of activity, ii) increasing the size of the C-3 and C-5 alkyl ester substituents decreases potency and iii) a C-4 1-oxido-4-pyridinyl substituent abolishes activity. The most active, and equipotent C-4 4-quinolinyl 6a and 8-quinolinyl 6b analogs, were approximately 8-fold less potent calcium channel antagonists than the reference drug nifedipine.

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### Introduction.

Vasculature-selective Hantzsch 1,4-dihydropyridine calcium channel antagonists, that close the L-type voltage dependent calcium channel, such as felodipine (1) exhibit a minimal inotropic effect and are useful for the treatment of hypertension and vasospasm [1]. Structure-activity relationships for dialkyl 1,4-dihydro-2,6-dimethyl-4-aryl-3,5pyridinedicarboxylates indicate that the nature and position of aryl ring substituents are determinants of calcium channel antagonist activity where the potency order is generally  $ortho \ge meta >> para$  [2,3]. Changes in the substitution pattern at the C-3, C-4 and C-5 positions of the 1,4-dihydropyridine system alter potency, tissue selectivity [4,5], and the conformation (degree of ring pucker) of the 1,4-dihydropyridine ring, which correlated well with activity. The most potent compounds show the smallest degree of ring distortion from planarity. Hantzsch 1,4-dihydropyridines, in the solid state, exist in a boat conformation where the C-4 substituted-phenyl ring is perpendicular (pseudoaxial) to the 1,4-dihydropyridine ring. Strain due to non-bonded interactions between the C-3, C-4 and C-5 1,4-dihydropyridine substituents is relieved most notably by puckering of the 1,4-dihydropyridine ring and distortion of the bond angles about C-4 [3,6,7].

In an earlier study, we showed that a 4-(pyridinyl) substituent 2 is bioisosteric with a 4-(nitrophenyl) substituent on a 1,4-dihydropyridine ring where *ortho-*, *meta-* and *para-*nitrophenyl are bioisosteric with 2-pyridinyl, 3-pyridinyl, and 4-pyridinyl, respectively [8]. These results indicate that the position of the pyridinyl nitrogen free electron-pair, and/or charge distribution [9,10] in the pyridinyl ring, may be capable of electrostatic binding to the  $\alpha_1$ -subunit binding site of the L-type calcium channel receptor [11] and ultimately a determinant of calcium channel antagonist activity. As part of our on-going program to study the structure-function relationship of calcium channels, we now report the syntheses and *in vitro* calcium channel antagonist

activities for dialkyl 1,4-dihydro-2,6-dimethyl-4-(quino-linyl)pyridine-3,5-dicarboxylates 6a-e, and the related C-4 1-oxido-4-pyridinyl analog 6f.

## Chemistry.

The dialkyl 1,4-dihydro-2,6-dimethyl-4-(heteroaryl)-pyridine-3,5-dicarboxylates 6a-f were synthesized using a three-component modified Hantzsch reaction [12,13]. Accordingly, condensation of the respective heteroaryl-carboxaldehyde 3a, 3b or 3c with the alkyl acetoacetate 4a, 4b or 4c and an alkyl 3-aminocrotonate 5a or 5b in ethanol at reflux afforded the respective 1,4-dihydropyridine product 6a-f in 28-86% yield as illustrated in Scheme 1. This modified Hantzsch reaction provides superior yields of unsymmetrical 1,4-dihydropyridine compounds such as 6c and 6d.

## Biological Results.

The *in vitro* calcium channel antagonist activities of compounds **6a-f**, determined as the molar concentration of test compound causing a 50% decrease in the slow component, or tonic contractile response, (IC<sub>50</sub>  $\pm$  SEM(standard error means), n = 3) in guinea pig ileum longitudinal smooth muscle induced by the muscarinic agonist carbachol (1.6 x 10<sup>-7</sup> M), were determined graphically from the dose-response curves [14]. The results are summarized in Table 1.

A comparison of the activities of the equipotent 4-quinolinyl 6a and 8-quinolinyl 6b isomers indicates that

Scheme 1 [a]

NHet=
$$C-H$$
 +  $CH_3-C-CH_2-C-O-R^1$  +  $CH_3-C-CH_2-C-O-R^2$  i  $R^2O_2C$   $R^2O_2C$ 

[a] Reagents and conditions: i, EtOH, reflux, 18-48 hours.

the position of the quinolinyl nitrogen is not a determinant of calcium channel antagonist activity. In the 8-quinolinyl series of compounds 6b-e, increasing the size of the alkyl ester substituents decreases potency [6b  $(R^1 = R^2 = Et) > 6e (R^1 = R^2 = i-Pr); 6b (R^1 = R^2 = Et) >$ **6d** ( $R^1 = Et$ ,  $R^2 = i$ -Pr)]. These results indicate that small R1 and R2 alkyl ester substituents such as Et, or preferably Me which would be expected to reduce the nonbonded interactions between the C-3, C-4 and C-5 substituents, should be selected when a large C-4 substituent such as 8-quinolinyl is present on the 1,4-dihydropyridine ring system. This structure-activity relationship differs from previous results for compounds 2 possessing a C-4 2-, 3- or 4-pyridinyl substituent where increasing the size of the alkyl ester substituents generally increased potency [8]. The C-4 8-quinolinyl ring system (6b,  $IC_{50} = 1.20 \text{ x}$ 10-7 M) is bioisosteric with a C-4 2-pyridinyl substituent present in 2 (R<sup>1</sup> = R<sup>2</sup> = Et, IC<sub>50</sub> = 1.77 x  $10^{-7}$  M) [8].

Table 1

Calcium Channel Antagonist Activities for Dialkyl 1,4-Dihydro-2,6dimethyl-4-(heteroaryl)pyridine-3,5-dicarboxylates 6a-f

Compound	R <sup>1</sup>	R <sup>2</sup>	Heterocyclic Substitutent	Calcium channel antagonist act:ivity IC <sub>50</sub> (M) [a]
6a	Et	Et	4-quinolinyl	$1.21 \pm 0.11 \times 10^{-7}$ (3)
6Ь	Et	Et	8-quinolinyl	$1.20 \pm 0.12 \times 10^{-7}$ (3)
6c	Me	i-Pr	8-quinolinyl	$5.23 \pm 0.17 \times 10^{-6}$ (3)
6d	Et	i-Pr	8-guinolinyl	$6.75 \pm 0.28 \times 10^{-5}$ (3)
6e	i-Pr	i-Pr	8-quinolinyl	$3.23 \pm 0.48 \times 10^{-5}$ (3)
6f	Et	Et	1-oxido- 4-pyridinyl	$> 1.0 \times 10^{-4} (3)$
Nifedipine [b]				$1.43 \pm 0.19 \times 10^{-8}$ (8)

[a] The molar concentration of antagonist test compound causing a 50% decrease in the slow component, or tonic contractile response, (IC<sub>50</sub>  $\pm$  SEM) in guinea pig ileum longitudinal smooth muscle by the muscarinic agonist carbachol (1.6 x 10<sup>-7</sup> M) was determined graphically from the dose-response curves. The number of experiments is shown in brackets. [b] Reference calcium channel antagonist drug (Adalat®).

Replacement of the C-4 4-quinolinyl substituent present in 6a (IC<sub>50</sub> = 1.21 x  $10^{-7}$  M) by a 1-oxido-4-pyridinyl substituent as in 6f (IC<sub>50</sub> = > 1.0 x  $10^{-4}$  M) abolished calcium channel antagonist activity. The most active 4-quinolinyl 6a and 8-quinolinyl 6b compounds were approximately 8-fold less potent calcium channel antagonists than the reference drug nifedipine.

## **EXPERIMENTAL**

Melting points were determined using a Thomas-Hoover capilliary apparatus and are uncorrected. Nuclear magnetic resonance resonance spectra (1H nmr) were recorded on a Bruker AM-300 spectrometer. The assignment of exchangeable protons (NH) was confirmed by the addition of deuterium oxide. Infrared spectra were acquired using a Nicolet 5DX-FT spectrometer. Silica gel column chromatography was carried out using Merck 7734 (60-200 mesh) silica gel. 4-Quinolinecarboxaldehyde (3a), 1-oxido-4-pyridinecarboxaldehyde (3c), ethyl acetoacetate (4a), methyl acetoacetate (4c), ethyl 3-aminocrotonate (5a) and isopropyl 3-aminocrotonate (5b) were purchased from the Aldrich Chemical Co. Isopropyl acetoacetate (4b) was purchased from the Lancaster Chemical Co. 8-Quinolinecarboxaldehyde (3b) was synthesized by the selenium dioxide oxidation of 8-methylquinoline (purchased from the Aldrich Chemical Co.) using a literature procedure [15].

General Procedure for the Synthesis of Dialkyl 1,4-Dihydro-2,6-dimethyl-4-(heteroaryl)pyridine-3,5-dicarboxylates 6a-f.

A mixture of the heteroarylcarboxaldehyde 3a, 3b or 3c (1 mmole), the alkyl acetoacetate 4a, 4b or 4c (1 mmole) and the alkyl 3-aminocrotonate 5a or 5b (1 mmole) in ethanol (50 ml) was heated at reflux until the reaction had gone to completion, 18-48 hours. Removal of the solvent *in vacuo*, purification of the residue obtained by silica gel column chromatography using ethyl acetate-hexane (2:1, v/v for 6a-d and 6a-f; 1:4, v/v for 6e) as eluent, and then recrystallization of the product from ethyl acetate-hexane (1:1, v/v) yielded the respective product 6a-f. The percentage yield, mp, ir and <sup>1</sup>H nmr spectral data for 6a-f are reported below.

Diethyl 1,4-Dibydro-2,6-dimethyl-4-(4-quinolinyl)pyridine-3,5-dicarboxylate (6a).

This compound was obtained in a yield of 78%, mp 203-204°; ir (potassium bromide): v 3182 (NH), 1691 (CO<sub>2</sub>) cm<sup>-1</sup>; <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  0.95 (t, J = 7.1 Hz, 6H, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 2.38 (s, 6H, 1,4-dihydropyridine C-2 and C-6 Me's), 3.90 (q, J = 7.1 Hz, 4H, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 5.84 (s, 1H, 1,4-dihydropyridine H-4), 6.47 (s, 1H, NH), 7.47 (d, J<sub>2,3</sub> = 4.8 Hz, 1H, quinolyl H-3), 7.59 (ddd, J<sub>5,6</sub> = J<sub>6,7</sub> = 8.0, J<sub>6,8</sub> = 1.0 Hz, 1H, quinolyl H-6), 7.70 (ddd, J<sub>6,7</sub> = J<sub>7,8</sub> = 8.0, J<sub>5,7</sub> = 1.0 Hz, 1H, quinolyl H-7), 8.09 (dd, J<sub>5,6</sub> = 8.0, J<sub>5,7</sub> = 1.0 Hz, 1H, quinolyl H-5), 8.64 (d, J<sub>7,8</sub> = 8.0 Hz, 1H, quinolyl H-8), 8.80 (d, J<sub>2,3</sub> = 4.8 Hz, 1H, quinolyl H-2).

Anal. Calcd. for C<sub>22</sub>H<sub>24</sub>N<sub>2</sub>O<sub>4</sub>: C, 69.46; H, 6.36; N, 7.36. Found: C, 69.24; H, 6.35; N, 7.41.

Diethyl 1,4-Dihydro-2,6-dimethyl-4-(8-quinolinyl)pyridine-3,5-dicarboxylate (6b).

This compound was obtained in a yield of 31%, mp 160-161°; ir (potassium bromide): v 3188 (NH), 1698 (CO<sub>2</sub>) cm<sup>-1</sup>; <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  0.80 (t, J = 7.1 Hz, 6H, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 2.26 (s, 6H, 1,4-dihydropyridine C-2 and C-6 Me's), 3.88 (q, J = 7.1 Hz, 4H, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 5.72 (s, 1H, 1,4-dihydropyridine H-4), 6.25 (s, 1H, NH), 7.29 (dd, J<sub>2,3</sub> = 5.0, J<sub>3,4</sub> = 8.0 Hz, 1H, quinolyl H-3), 7.40 (t, J<sub>5,6</sub> = J<sub>6,7</sub> = 7.5 Hz, 1H, quinolyl H-6), 7.56-7.67 (m, 2H, quinolyl H-5 and H-7), 8.13 (dd, J<sub>3,4</sub> = 8.0, J<sub>2,4</sub> = 1.5 Hz, 1H, quinolyl H-4), 8.93 (dd, J<sub>2,3</sub> = 5.0, J<sub>2,4</sub> = 1.5 Hz, 1H, quinolyl H-2).

Anal. Calcd. for C<sub>22</sub>H<sub>24</sub>N<sub>2</sub>O<sub>4</sub>: C, 69.46; H, 6.36; N, 7.36. Found: C, 69.13; H, 6.60; N, 7.18.

3-Isopropyl 5-Methyl 1,4-Dihydro-2,6-dimethyl-4-(8-quino-linyl)pyridine-3,5-dicarboxylate (6c).

This compound was obtained in a yield of 36%, mp 192-193°; ir (potassium bromide): v 3183 (NH), 1704 (CO<sub>2</sub>) cm<sup>-1</sup>; <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  0.56 and 1.08 (two d, J = 6.2 Hz, 3H each, CHMe<sub>2</sub>), 2.26 and 2.28 (two s, 3H each, 1,4-dihydropyridine C-2 and C-6 Me's), 3.38 (s, 3H, CO<sub>2</sub>CH<sub>3</sub>), 4.63 (septet, J = 6.2 Hz, 1H, CHMe<sub>2</sub>), 5.59 (s, 1H, 1,4-dihydropyridine H-4), 6.23 (s, 1H, NH), 7.30 (dd, J<sub>2,3</sub> = 5.0, J<sub>3,4</sub> = 8.0 Hz, 1H, quinolyl H-3), 7.41 (t, J<sub>5,6</sub> = J<sub>6,7</sub> = 8.0 Hz, 1H, quinolyl H-6), 7.56-7.66 (m, 2H, quinolyl H-5 and H-7), 8.04 (dd, J<sub>3,4</sub> = 8.0, J<sub>2,4</sub> = 1.5 Hz, 1H, quinolyl H-4), 8.95 (dd, J<sub>2,3</sub> = 5.0, J<sub>2,4</sub> = 1.5 Hz, 1H, quinolyl H-2).

Anal. Calcd. for C<sub>22</sub>H<sub>24</sub>N<sub>2</sub>O<sub>4</sub>: C, 69.46; H, 6.36; N, 7.36. Found: C, 69.12; H, 6.37; N, 7.25.

3-Isopropyl 5-Ethyl 1,4-Dihydro-2,6-dimethyl-4-(8-quinolinyl)-pyridine-3,5-dicarboxylate (6d).

This compound was obtained in a yield of 28%, mp 96-97°; ir (potassium bromide): v 3217 (NH), 1696 (CO<sub>2</sub>) cm<sup>-1</sup>; <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  0.50-0.53 (m, 6H total, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub> and CO<sub>2</sub>CHCH<sub>3</sub>), 0.73 (d, J = 6.1 Hz, 3H, CO<sub>2</sub>CHCH<sub>3</sub>), 2.78 (s, 6H, 1,4-dihydropyridine C-2 and C-6 Me's), 3.73 (q, J = 7.1 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 4.66 (septet, J = 6.1 Hz, 1H, CHMe<sub>2</sub>), 5.59 (s, 1H, 1,4-dihydropyridine H-4), 6.23 (s, 1H, NH), 7.43 (dd, J<sub>2,3</sub> = 4.3, J<sub>3,4</sub> = 8.3 Hz, 1H, quinolyl H-3), 7.55-7.63 (m, 2H, quinolyl H-5 and H-7), 7.85-7.95 (m, 1H, quinolyl H-6), 8.20 (dd, J<sub>3,4</sub> = 8.3, J<sub>2,4</sub> = 1.7 Hz, 1H, quinolyl H-4), 8.91 (dd, J<sub>2,3</sub> = 4.3, J<sub>2,4</sub> = 1.7 Hz, 1H, quinolyl H-2).

Anal. Calcd. for C<sub>23</sub>H<sub>26</sub>N<sub>2</sub>O<sub>4</sub>: C, 70.03; H, 6.64; N, 7.10. Found: C, 70.10; H, 6.31; N, 7.00.

Diisopropyl 1,4-Dihydro-2,6-dimethyl-4-(8-quinolinyl)pyridine-3.5-dicarboxylate (6e).

This compound was obtained in a yield of 40%, mp 162-163°; ir (potassium bromide): v 3201 (NH), 1698 (CO<sub>2</sub>) cm<sup>-1</sup>; <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  0.50 and 0.71 (two d, 6H each, CHMe<sub>2</sub>), 2.71 (s, 6H, 1,4-dihydropyridine C-2 and C-6 Me's), 4.64 (septet, J = 6.1 Hz, 1H, CHMe<sub>2</sub>), 5.59 (s, 1H, 1,4-dihydropyridine H-4), 6.23 (s, 1H, NH), 7.41 (d, J<sub>2,3</sub> = 4.2 Hz, 1H, quinolyl H-3), 7.56 (d, J<sub>5,6</sub> = J<sub>6,7</sub> = 8.1 Hz, 2H, quinolyl H-5 and H-7), 7.88 (t, J<sub>5,6</sub> = J<sub>6,7</sub> = 8.1 Hz, 1H, quinolyl H-6), 8.18 (dd, J<sub>3,4</sub> = 8.3, J<sub>2,4</sub> = 1.8 Hz, 1H, quinolyl H-4), 8.90 (dd, J<sub>2,3</sub> = 4.2, J<sub>2,4</sub> = 1.8 Hz, 1H, quinolyl H-2).

Anal. Calcd. for C<sub>24</sub>H<sub>28</sub>N<sub>2</sub>O<sub>4</sub>: C, 70.57; H, 6.91; N, 6.86. Found: C, 70.70; H, 6.58; N, 6.77.

Diethyl 1,4-Dihydro-2,6-dimethyl-4-(1-oxido-4-pyridinyl)-pyridine-3,5-dicarboxylate (6f).

This compound was obtained in a yield of 86%, mp 218-220°; ir (potassium bromide): v 3172 (NH), 1688 (CO<sub>2</sub>), 1231 (NO) cm<sup>-1</sup>; <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  1.23 (t, J = 7.1 Hz, 6H, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 2.38 (s, 6H, 1,4-dihydropyridine C-2 and C-6 Me's), 4.12 (q, J = 7.1 Hz, 4H, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 5.02 (s, 1H, 1,4-dihydropyridine H-4), 7.21 (s, 1H, NH), 7.31 (d,  $J_{2,3}$  =  $J_{5,6}$  = 7.0 Hz, 2H, 1-oxido-4-pyridinyl H-3 and H-5), 8.10 (dd,  $J_{2,3}$  =  $J_{5,6}$  = 7.0,  $J_{2,6}$  = 1.5 Hz, 2H, 1-oxido-4-pyridinyl H-2 and H-6).

Anal. Calcd. for C<sub>18</sub>H<sub>22</sub>N<sub>2</sub>O<sub>5</sub>•1/4 H<sub>2</sub>O: C, 61.62; H, 6.61; N, 7.98. Found: C, 61.77; H, 6.35; N, 7.94.

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